

WHAT HAPPENS DURING A **DUI ARREST?**



A **DUI (Driving Under the Influence) arrest** usually begins with an **ordinary traffic stop**. There are specific things that the officer is looking for, and how you handle the traffic stop can help or hurt your **DUI defense attorney** when he or she is trying to get your charges dismissed or reduced.

Probable Cause

Most **DUI arrests** begin with some type of **traffic violation**, such as speeding, following too closely, or weaving. In some cases your **DUI lawyer** will be able to challenge the officer's probable cause for the initial traffic stop. Your **DUI lawyer** may also be able to challenge the officer's determination that there was probable cause to believe that you were intoxicated.

In most cases, the officer will put in their report that they arrested the car of alcohol, that you were glassy-eyed or had bloodshot eyes, that your speech was slurred, or that you fumbled when looking for your license or insurance information.

Field Sobriety Tests

Once the officer believes that you have been drinking, the officer will usually offer a series of **sobriety tests**, beginning with the **standardized field sobriety tests on the roadside** and ending with a **breathalyzer or blood test** after the arrest has been made.

Whether you agree to take these tests and your performance on them will be the most critical evidence that your **DUI defense attorney** has to overcome in your case, and can determine whether your case will ultimately be a plea to reduced charges, a dismissal, or an acquittal at trial.

Hiring a Lawyer

Hiring a good **DUI lawyer** as quickly as possible is crucial if you are charged with **DUI**. The sooner you speak with an attorney, the sooner he or she can start preparing your case for a successful outcome. If you are arrested, reserve your right to remain silent and don't speak with anyone but your lawyer.



Dui How To Prevent Your Arrest

Yi-Tong Ma



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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) + \text{C}_2\text{H}_2(\text{g}) \rightarrow \text{C}_2\text{H}_6(\text{g}) + 2 \text{ H}_2(\text{g}) + \text{C}_2\text{H}_4(\text{g})$ $\Delta h = 94.5 \text{ kJ}$
 $\text{H}_2\text{O}(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l})$ $\Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{g})$ $\Delta h = 283 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g})$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 2 \text{ H}_2\text{O}(\text{l})$ $\Delta h = 1411 \text{ kJ/mole}$
 $\text{C}_2\text{H}_6(\text{g}) + 7/2 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{l})$ $\Delta h = 1560 \text{ kJ/mole}$
 $\text{H}_2(\text{g}) + 1/2 \text{ O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l})$ $\Delta h = 285 \text{ kJ/mole}$

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