



# Double Pink

**Daniel F McAuley**



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mol ko 2 and 0 mol h 2 o what is the limiting reactant

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$$2\text{Cu} + 3\text{AgNO}_3 \rightarrow 2\text{Ag} + \text{Cu}(\text{NO}_3)_2 + 3\text{AgNO}_3$$

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80 g  $\text{I}_2\text{O}_5$  1 mol  $\text{I}_2\text{O}_5$  1 mol  $\text{I}_2$  xs 1 333.8 g  $\text{I}_2\text{O}_5$  1 mol  $\text{I}_2\text{O}_5$  28 g  $\text{CO}$  1 mol  $\text{CO}$

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