

JULY 2020						
S	M	T	W	T	F	S
			1	2	3	4
5	6	7	8	9	10	11
12	13	14	15	16	17	18
19	20	21	22	23	24	25
26	27	28	29	30	31	

JULY
 3 – HOLIDAY - 4th of July
 22 – Counselor Start Date
 27, 28, 29 – New Teacher Academy
 30 – New teacher Work Day
 31 – All Teachers Return
 31 – Full Site Professional Development

AUGUST 2020						
S	M	T	W	T	F	S
						1
2	3	4	5	6	7	8
9	10	11	12	13	14	15
16	17	18	19	20	21	22
23	24	25	26	27	28	29
30	31					

AUGUST
 3 – Back to School Kick Off 1/2 Day Professional Development
 3 – 1/2 Day Teacher Work Day
 4 – Teacher Work Day
 5 – 1st Student Day

SEPTEMBER 2020						
S	M	T	W	T	F	S
		1	2	3	4	5
6	7	8	9	10	11	12
13	14	15	16	17	18	19
20	21	22	23	24	25	26
27	28	29	30			

SEPTEMBER
 7 – HOLIDAY - Labor Day

OCTOBER 2020						
S	M	T	W	T	F	S
				1	2	3
4	5	6	7	8	9	10
11	12	13	14	15	16	17
18	19	20	21	22	23	24
25	26	27	28	29	30	31

OCTOBER
 2 – End of 1st Quarter
 5-9 – Fall Break

NOVEMBER 2020						
S	M	T	W	T	F	S
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30					

NOVEMBER
 11 – HOLIDAY - Veterans Day
 23-27 – Thanksgiving Break (Cert. & Class. w/less than 12 months)
 25-27 – HOLIDAY - Thanksgiving

DECEMBER 2020						
S	M	T	W	T	F	S
		1	2	3	4	5
6	7	8	9	10	11	12
13	14	15	16	17	18	19
20	21	22	23	24	25	26
27	28	29	30	31		

DECEMBER
 18 – End of 2nd Quarter (34 days)
 End of 1st Semester (36 Days)
 All secondary Grades due
 21-31 – Winter Break (Cert. & Class. w/less than 12 months)
 24, 25 – HOLIDAY - Winter Break
 29-31 – Non-Work Days (12 Month-Classified Employees)

	Holiday Schools Closed
	Holiday District Office and Schools Closed
	Professional Development Days
	Teacher Work Days
	New Teacher Academy

JANUARY						
S	M	T	W	T	F	S
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30	31				

JANUARY
 1 – HOLIDAY - New Year's
 4 – Professional Development Day
 5 – Teacher Work Day
 6 – Students Return
 18 – HOLIDAY - Martin Luther King Day

FEBRUARY						
S	M	T	W	T	F	S
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30					

FEBRUARY
 1 – HOLIDAY – Observance of Lincoln's Birthday
 15 – HOLIDAY - President's Day

MARCH						
S	M	T	W	T	F	S
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30	31				

MARCH
 12 – End of 3rd Quarter (45 Days)
 22-31 – Spring Break (Cert. & Class. w/less than 12 months)

JANUARY 2021						
S	M	T	W	T	F	S
3	4	5	6	7	8	9
10	11	12	13	14	15	16
17	18	19	20	21	22	23
24	25	26	27	28	29	30
31						

JANUARY 2021
 (17 Days Taught)

FEBRUARY 2021						
S	M	T	W	T	F	S
		1	2	3	4	5
6	7	8	9	10	11	12
13	14	15	16	17	18	19
20	21	22	23	24	25	26
27	28	29	30			

(18 Days Taught)

MARCH 2021						
S	M	T	W	T	F	S
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30	31				

MARCH 2021
 (13 Days Taught)

APRIL 2021						
S	M	T	W	T	F	S
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30					

(20 Days Taught)

MAY 2021						
S	M	T	W	T	F	S
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30	31				

MAY 2021
 (20 Days Taught)

JUNE 2021						
S	M	T	W	T	F	S
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30					

(4 Days Taught)

	First Student Day (August 5)
	Students return from Winter Break (January 6)
	Last Student Day (June 4)
	Student Days – 180 days Returning Teacher Days – 185 days
	New Teacher Days – 187 Days

Ball State University Event Calendar 2014

Mark William Padilla



Ball State University Event Calendar 2014:

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Human Ecology, 2014

Baseball Meets the Law Ed Edmonds, Frank G. Houdek, 2017-03-04 Baseball and law have intersected since the primordial days In 1791 a Pittsfield Massachusetts ordinance prohibited ball playing near the town's meeting house Ball games on Sundays were barred by a Pennsylvania statute in 1794 In 2015 a federal court held that baseball's exemption from antitrust laws applied to franchise relocations Another court overturned the conviction of Barry Bonds for obstruction of justice A third denied a request by rooftop entrepreneurs to enjoin the construction of a massive video screen at Wrigley Field This exhaustive chronology traces the effects the law has had on the national pastime both pro and con on and off the field from the use of copyright to protect not only equipment but also Take Me Out to the Ball Game to frequent litigation between players and owners over contracts and the reserve clause The stories of lawyers like Kenesaw Mountain Landis and Branch Rickey are entertainingly instructive

Classical Myth in Alfred Hitchcock's Wrong Man and Grace Kelly Films Mark William Padilla, 2018-12-12 Mark Padilla's classical reception readings of Alfred Hitchcock features some of the director's most loved and important films and demonstrates how they are informed by the educational and cultural classicism of the director's formative years The six close readings begin with discussions of the production histories so as to theorize and clarify how classicism could and did enter the projects Exploration of the films through a classical lens creates the opportunity to explore new themes and ideological investments The result is a further appreciation

of both the engine of the director's storytelling creativity and the expressionism of classicism especially Greek myth and art in British and American modernism The analysis organizes the material into two triptychs one focused on the three films sharing a wrong man pattern wrongly accused man goes on the run to clear himself the other treating the films starring the actress Grace Kelly Chapter One on *The 39 Steps* 1935 finds the origins of the wrong man plot in early 20th century British classicism and demonstrates that the movie utilizes motifs of Homer's *Odyssey* Chapter Two on *Saboteur* 1942 theorizes the impact of the director's memories of the formalism and myths associated with the Parthenon sculptures housed in the British Museum Chapter Three on *North by Northwest* participates in the myths of the hero Oedipus as associated with early Greek epic Freud Nietzsche and Sophocles Chapter Four on *Dial M for Murder* 1954 returns to Homer's *Odyssey* in the interpretive use of the lay of Demodocus a story about the sexual triangle of Hephaestus Aphrodite and Ares Chapter Five on *Rear Window* 1954 finds its narrative archetype in The Homeric Hymn to Aphrodite the erotic theme of Sirius the Dog Star also marks the film Chapter Six on *To Catch a Thief* 1955 offers the opportunity to break from mythic analogues and to consider the film's philosophical resonances Plato and Epicurus in the context of motifs coalesced around the god Dionysus Bacchus

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Library Journal, 1972-10 **Journal of the American Veterinary Medical Association** American Veterinary Medical Association, 2010 **Human Resources Report**, 1997 Commonwealth Yearbook Nexus Strategic Publishers, 2014-07-28

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web jan 26 2016 the solubility product or ksp is a way of showing how effective a chemical is at dissolving into an aqueous solution the higher the ksp the more the chemical will dissolve into that solvent for this lab calcium hydroxide is created then ionized $Ca(OH)_2$ when calcium is put in water reacts as follows $Ca(OH)_2 \rightleftharpoons Ca^{2+} + 2OH^{-}$

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web k sp of calcium hydroxide 1 author j m mccormick last update ocober 13 2013 introduction an equilibrium constant k is related to ΔG for a process through eqn 1 and as long as ΔH and ΔS are independent of temperature eqn 1 can be combined with eqn 2 to determine ΔH and ΔS from the temperature dependence of k

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web sep 18 2020 in this experiment we will titrate a saturated calcium hydroxide solution with a standardized hcl solution we will determine the hydroxide ion concentration for the saturated calcium hydroxide

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web your primary objective in this experiment is to test a saturated solution of calcium hydroxide and use your observations and measurements to calculate the ksp of the compound you will do this by titrating the prepared Ca(OH)_2 solution with a standard hydrochloric acid solution

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web in today's experiment you will determine the solubility product ksp of calcium hydroxide Ca(OH)_2 by measuring the concentration of Ca(OH)_2 in a saturated solution calcium hydroxide is a sparingly soluble salt that dissolves according to the following reaction the solubility product expression for this reaction is

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web expert verified chem 182 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water a saturated aqueous solution of Ca(OH)_2 is represented in equation form as shown below $\text{Ca(OH)}_2(\text{s}) \rightleftharpoons \text{Ca}^{2+}(\text{aq}) + 2\text{OH}^{-}(\text{aq})$ the solubility product expression describes in mathematical terms the equilibrium

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web ksp for calcium hydroxide calcium hydroxide is a soft white caustic powder used in making mortar cements calcium salts paints and petrochemicals it is also used in saltwater aquaria to make up kalkwasser limewater solutions for reef tanks and is used as a pH regulating agent notice that calcium hydroxide is divalent and thus has twice

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web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution $\text{Ca(OH)}_2 \rightarrow \text{Ca}^{2+} + 2\text{OH}^{-}$ $5 \times 10^{-2} \text{ M}$ calcium ions $5 \times 10^{-2} \text{ M}$ $2 \times 10^{-2} \text{ M}$ hydroxide ions calculate calcium hydroxide ksp using the concentrations of calcium ions and hydroxide ions

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web madison bowe elana keller april 15 2014 introduction the objective of this lab was to measure the ksp of a slightly soluble basic salt and how solubility changes with temperature ksp is another form of equilibrium expression that deals with solubility the reaction that will take place is $\text{Ca(OH)}_2(\text{s}) \rightleftharpoons \text{Ca}^{2+}(\text{aq}) + 2\text{OH}^{-}(\text{aq})$

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web question experiment 8 determining ksp for calcium hydroxide post lab report sheet data table 2 points trial 1 trial 2 hci molarity 0.050 M volume of filtered Ca(OH)_2 solution ml 5.00 5.00 3.71 13.75 equivalence point volume ml OH⁻ in saturated Ca(OH)_2 M Ca²⁺ in saturated Ca(OH)_2 M ksp of Ca(OH)_2 average ksp of Ca(OH)_2

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web the chemical equation used was $\text{Ca(OH)}_2(\text{s}) \rightleftharpoons \text{Ca}^{2+}(\text{aq}) + 2\text{OH}^{-}(\text{aq})$ the solubility product was determined to be $K_{sp} = [\text{Ca}^{2+}][\text{OH}^{-}]^2$ from the

titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

18 1 solubility product constant ksp chemistry libretxts - Feb 01 2023

web jul 12 2023 because the concentration of a pure solid such as Ca_3PO_4 is a constant it does not appear explicitly in the equilibrium constant expression the equilibrium constant expression for the dissolution of calcium phosphate is therefore $K_{\text{sp}} = [\text{Ca}^{2+}]^3 [\text{PO}_4^{3-}]^2$

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web lab report 8 determinin of calcium hydroxide the purpose of this was to titrat e a saturated $\text{Ca}(\text{OH})_2$ soultion with a st andard hcl solution and find the ksp or the oh concentration was f ound and then the concentration was found and u for this experiment a titration of calcium hydroxide with hydrochlo

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web question determining the ksp of calciumhydroxide $\text{Ca}(\text{OH})_2$ s Ca^{2+} aq 2OH^- aq for this experiment the titration of calciumhydroxide with hydrochloric acid wasinvestigated determining the ksp of calcium hydroxide $\text{Ca}(\text{OH})_2$ s Ca^{2+} aq 2OH^- aq for this experiment the titration of calcium

solved chem 182 experiment 8 determining the ksp of calcium chegg - Sep 27 2022

web question chem 182 experiment 8 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water solution of $\text{Ca}(\text{OH})_2$ is represented in equation form lution of $\text{Ca}(\text{OH})_2$ on s nronic solid that is sparingly soluble in water a saturated aqueous as shown below

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web therefore it was determined through error calculations and ease of use that the most reliable valid method to use to get the closest ksp value to the theoretical ksp for $\text{Ca}(\text{OH})_2$ was the determination through ph method 2 introduction calcium hydroxide is a strong base soluble in water

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web plasma membrane homework symbiosis practice worksheet benzoin condensation grade a determining the ksp of calcium hydroxide after reading the lab complete items a b c and d title purpose chemicals and equipment and summary of procedure as described on page 10 of exp 1 on an 8 1 2 x 11 sheet of paper

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web question lab 9 determination of the solubility product ksp of calcium hydroxide introduction ionic compounds that are classified as insoluble based on solubility rules are actually slightly solubla each of these insoluble compounds actually dissolves to

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we will filter the solid calcium hydroxide from the solution and then titrate the dissolved calcium hydroxide in solution with a known concentration of hydrochloric acid using titrimetric calculations we will determine the Ca^{2+} and OH^- concentrations in solution

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we calculate the ksp for calcium hydroxide $\text{Ca}^{2+} \text{OH}^-$ 2.0×10^{-2} ksp 5×10^{-4} find the accepted value of the ksp for calcium hydroxide and compare it with your value discuss the discrepancy and suggest possible sources of experimental error
accepted value for ksp for sodium hydroxide is 5×10^{-6}