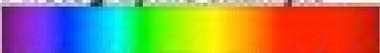
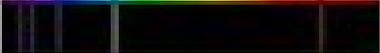


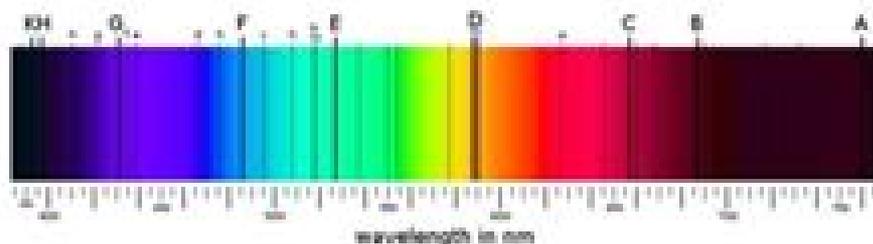
Spectral Analysis

Worksheet

Types of spectra:

Type of spectrum	Photographic example
Continuous (or continuum)	
Absorption (dark line)	
Emission (bright line)	

Here is a spectrum with many of the known spectral lines.

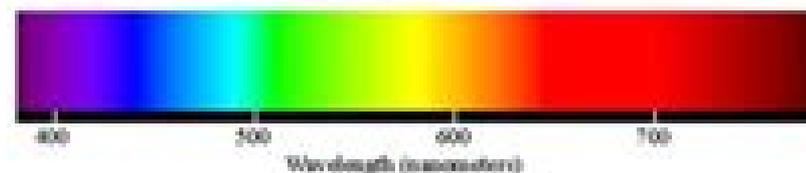


Designation	Element	Wavelength (nm)	Designation	Element	Wavelength (nm)
y	O ₂	844.765	v	Fe	485.763
Z	O ₂	822.694	w	H β	486.134
A	O ₂	779.379	u	Fe	466.814
B	O ₂	686.719	t	Fe	438.355
C	H α	656.281	Q	H γ	434.047
e	O ₂	627.641	G	Fe	430.790
D ₁	Na	589.592	G	Ca	430.774
D ₂	Na	589.595	h	H δ	430.175
D ₃ or d	H α	587.5622	H	Ca	396.847
e	H γ	544.813	K	Ca	395.348
E ₁	Fe	527.659	L	Fe	382.044
h ₁	Mg	518.342	N	Fe	358.123
h ₂	Mg	518.279	P	Zr	356.132
h ₃	Fe	514.891	T	Fe	352.108
h ₄	Fe	514.891	i	Si	299.444
h ₅	Mg	516.733			

- What colour is a photon of light with wavelength
 - 450 nm _____
 - 700 nm _____
 - 523 nm _____
 - 560 nm _____

- What colour is the emission of hydrogen-alpha, H α ?

- Draw the main spectrum lines for a star containing Iron and Hydrogen. (note: Some elements have more than one absorption line)



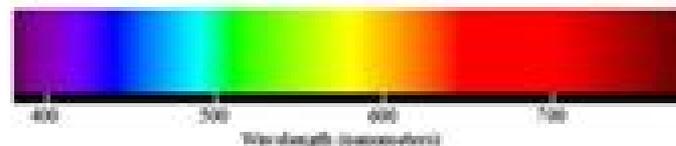
- Draw the Emission Lines on a spectrum for a Magnesium lamp. Insert a wavelength value.



- A Nebula of mainly Hydrogen and Mercury (unlikely) is between a star and the earth.

- What happens to the light as it passes through the nebula

- The Star is mainly hydrogen and Helium. Complete the spectrum



An Introduction To Spectral Analysis

Ying-Ying Zheng



An Introduction To Spectral Analysis:

The book delves into An Introduction To Spectral Analysis. An Introduction To Spectral Analysis is a crucial topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into An Introduction To Spectral Analysis, encompassing both the fundamentals and more intricate discussions.

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 - Chapter 5: Conclusion
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