

Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe a spontaneous redox reaction. (10 marks)

- a. Anode: _____
 - b. Cathode: _____
 - c. Anion: _____
 - d. Cation: _____
 - e. Half-cell reactions: Zn: _____
 - f. Spontaneous overall reaction: _____
 - g. Write overall balanced cell: _____
 - h. E_{cell}: _____
 - i. Write overall cell process: $\text{Zn} + \text{Cu}^{2+} \rightarrow \text{Zn}^{2+} + \text{Cu}$ (10 marks)
- Write a spontaneous redox reaction for copper & silver.

Voltaic Cells Study Guide

LP Steffe



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