

## 17.5

## HYDROGEN IONS AND ACIDITY

## Section Overview

## Objectives

1. Calculate concentrations based on the pH of a solution or calculate the pH of a solution based on concentration.
2. Calculate pH values and concentrations of acids and bases and the concentrations of their conjugate acids and bases.
3. Calculate the pH of weak acid solutions.

## Vocabulary

- |                   |                                       |                    |
|-------------------|---------------------------------------|--------------------|
| 1. autoionization | 2. conjugate acid                     | 3. pH              |
| 4. autoionization | 5. hydronium ion ( $H_3O^+$ )         | 6. acid ionization |
|                   | 7. acid ionization constant ( $K_a$ ) | 8. acid ionization |
|                   | 9. acid ionization                    | 9. $pH$            |

## Key Equations

- |  |  |
|--|--|
| 1. $[H_3O^+] = 10^{-pH}$ (Equation 17.1) | 2. $pH = -\log [H_3O^+]$ (Equation 17.2) |
| 3. $pH = -\log [H_3O^+]$ (Equation 17.3) | 4. $[H_3O^+] = 10^{-pH}$ (Equation 17.4) |

## Part A: Completion

Use the appropriate number to identify the relationship of the conjugate acid and base of the acid or base listed. Write the number in the space provided.

1. What product is the \_\_\_\_\_ of a weak hydrogen ion donor?
2. \_\_\_\_\_ is the conjugate base of a weak acid.
3. \_\_\_\_\_ is the conjugate acid of a weak base.
4. \_\_\_\_\_ is the conjugate base of a weak acid.
5. \_\_\_\_\_ is the conjugate acid of a weak base.
6. \_\_\_\_\_ is the conjugate acid of a weak base.
7. \_\_\_\_\_ is the conjugate base of a weak acid.
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9. \_\_\_\_\_ is the conjugate base of a weak acid.
10. \_\_\_\_\_ is the conjugate acid of a weak base.
11. \_\_\_\_\_ is the conjugate base of a weak acid.
12. \_\_\_\_\_ is the conjugate acid of a weak base.

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# 19 2 Section Review Hydrogen Ions And Acidity

**Michael Brown**



## **19 2 Section Review Hydrogen Ions And Acidity:**

## Enjoying the Song of Expression: An Mental Symphony within **19 2 Section Review Hydrogen Ions And Acidity**

In a global eaten by monitors and the ceaseless chatter of instantaneous connection, the melodic beauty and psychological symphony developed by the written word often disappear in to the back ground, eclipsed by the persistent sound and distractions that permeate our lives. However, nestled within the pages of **19 2 Section Review Hydrogen Ions And Acidity** a stunning literary prize filled with raw feelings, lies an immersive symphony waiting to be embraced. Constructed by a wonderful composer of language, that fascinating masterpiece conducts visitors on an emotional trip, well unraveling the hidden tunes and profound impact resonating within each carefully constructed phrase. Within the depths of the poignant assessment, we shall investigate the book is central harmonies, analyze its enthralling publishing style, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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