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 $4\text{S} + 2\text{Ba} + \text{aq} \rightarrow 4\text{S} + 2\text{Ba} + \text{aq} + \text{Ag}_2\text{SO}_4 + 2\text{Ag}_2\text{SO}_4$
 $2\text{Al} + \text{OH}^- + 3\text{S} + \text{Al} + 3\text{aq} + 3\text{OH}^- + \text{aq} + \text{Pb} + \text{OH}^- + \text{Cl} + \text{S} + \text{Pb} + 2\text{aq} + \text{OH}^- + \text{aq} + \text{Cl} + \text{aq} + \text{aq}$

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