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web toluene is oxidized to benzoic acid and a small amount of benzaldehyde the kinetics of the reactions monitored by uv vis spectrometry show that the initial reactions are first order in the concentrations of both  $\text{KMnO}_4$  and substrate no induction periods are observed

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web an improved kinetic model for the high temperature oxidation of toluene has been developed using previously established reaction mechanisms for benzene and toluene

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formation and elimination

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web may 15 2020 toluene 1 oxidation with kmno4 mechanism 2 benzoic acid c6h5cooh socl2 3 benzoyl chloride c6h5cocl  
lialh oc ch3 3 h 4 benzaldehyde c6h5cho ch2ohch2oh h step 1 toluene oxidation with kmno4 kmno4 is a strong oxidising  
agent which oxidise toluene to benzaldehyde in 1st step

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web may 1 2004 oxidation of toluene using aqueous potassium permanganate was studied under heterogeneous condition in  
the presence of hydrodynamic cavitation and compared with the results of the reaction under acoustic cavitation

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web oct 15 2014 1 answer sorted by 6 here are the three equations describing the reduction of manganese and concurrent  
oxidation of whatever substrate may be present under basic neutral and acidic conditions respectively mnx 7ox4x ex mnx  
6ox4x2 basic m n x 7 o x 4 x e x m n x 6 o x 4 x 2 b a s i c

**what is the mechanism of toluene oxidation by kmno4 reddit** - Jul 14 2023

web mar 15 2019 what is the mechanism of toluene oxidation by kmno4 it is quite simple reaction that kmno4 can oxidize  
benzylic hydrogen in toluene and synthesize benzoic acid however i cannot find exact mechanism of reaction in

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web description treatment of an alkylbenzene with potassium permanganate results in oxidation to give the benzoic acid  
notes the position directly adjacent to an aromatic group is called the benzylic position the reaction only works if there is a  
hydrogen attached to the carbon examples

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web may 1 2004 the oxidation of toluene by aq kmno 4 under hydrodynamic cavitation was taken as a model reaction and  
various parameters have been optimized the oxidation of toluene by aq kmno 4 gives benzoic acid scheme 1 kmno 4 in turn is  
reduced to mno 2 the reaction did not give any other byproduct