

## CHEMISTRY UNIT 1 STUDY GUIDE.

### What is Matter?

- Solid, Liquid, gas
- Anything that takes up space & has mass
- Mass is the amount of matter in an object.

### Types

- Substance- a particular kind of matter – pure- made up of elements and compounds
- Mixture- more than one kind of matter. 2+ elements and compounds

### Properties:

**Physical** – property that can be observed and measure without changing the substances.

**Chemical** – property that can only be observed by changing the type of substance.

**Extensive**- only depends on the amount of matter

Mass, weight, volume, length

**Intensive** – only depends on the type of matter, not the amount used to identify a substance.

Color, Odeur, melting/ freezing point.

**Ductility**- the ability to make copper into wires.

### States of Matter:

	Definite volume	Definite shape	Temperature increase	compressable
SOLID	Yes	Yes	Small expans	No
LIQUID	Yes	No Takes up shape of container	Small expans	No
GAS	No	No	Large expans	Yes



# Science Review Sheet Unit Freshman

**Johns Hopkins University. Summer  
Session**



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## **Science Review Sheet Unit Freshman** Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the energy of words has be more evident than ever. They have the ability to inspire, provoke, and ignite change. Such may be the essence of the book **Science Review Sheet Unit Freshman**, a literary masterpiece that delves deep into the significance of words and their affect our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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