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web write the oxidation and reduction reactions for each redox reaction the first one is done for you $23 \text{ Fe} + \text{Co} \rightarrow \text{Co} + 2 \text{ Fe}$
 oxidation $\text{Co} + 2 \text{e}^-$ reduction $\text{Fe} + 2 \text{e}^-$ $24 \text{ 3 Ag} + \text{Ni} \rightarrow 3 \text{ Ag} + \text{Ni}$ oxidation reduction $25 \text{ Cu} + \text{Pb} \rightarrow \text{Pb} + \text{Cu}$ oxidation reduction 26 O_2

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 $5 \text{ CO} + \text{I}_2 \rightarrow 5 \text{ CO}_2 + \text{I}_2$ $5 \text{ C} + \text{O}_2 \rightarrow 5 \text{ CO}_2$

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