

Chemistry Final Exam Review

Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- _____ 1. The study of chemicals that, in general, do not contain carbon is traditionally called what type of chemistry?
a. bio
b. inorganic
c. physical
d. analytical
- _____ 2. Which of the following is NOT an example of matter?
a. air
b. heat
c. smoke
d. water vapor
- _____ 3. A golf ball has more mass than a tennis ball because it _____.
a. takes up more space
b. contains more matter
c. contains different kinds of matter
d. has a definite composition
- _____ 4. An example of an extensive property of matter is _____.
a. temperature
b. pressure
c. mass
d. hardness
- _____ 5. All of the following are physical properties of matter EXCEPT _____.
a. mass
b. color
c. melting point
d. ability to rust
- _____ 6. Which state of matter has a definite volume and takes the shape of its container?
a. solid
b. liquid
c. gas
d. both b and c
- _____ 7. Which of the following CANNOT be classified as a substance?
a. table salt
b. air
c. nitrogen
d. gold
- _____ 8. Which of the following is a heterogeneous mixture?
a. air
b. salt water
c. steel
d. soil
- _____ 9. Which of the following is a heterogeneous mixture?
a. vinegar in water
b. milk
c. oil and vinegar
d. air
- _____ 10. Separating a solid from a liquid by evaporating the liquid is called _____.
a. filtration
b. condensation
c. solution
d. distillation
- _____ 11. A substance that can be separated into two or more substances only by a chemical change is a(n) _____.
a. solution
b. element
c. mixture
d. compound
- _____ 12. The chemical symbol for iron is _____.
a. fe
b. FE
c. Fe
d. Ir

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Guillaume Favre



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Table of Contents 2nd Semester Biology Final Exam Review Packet

1. Understanding the eBook 2nd Semester Biology Final Exam Review Packet
 - The Rise of Digital Reading 2nd Semester Biology Final Exam Review Packet
 - Advantages of eBooks Over Traditional Books
2. Identifying 2nd Semester Biology Final Exam Review Packet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 2nd Semester Biology Final Exam Review Packet
 - User-Friendly Interface
4. Exploring eBook Recommendations from 2nd Semester Biology Final Exam Review Packet
 - Personalized Recommendations
 - 2nd Semester Biology Final Exam Review Packet User Reviews and Ratings
 - 2nd Semester Biology Final Exam Review Packet and Bestseller Lists

5. Accessing 2nd Semester Biology Final Exam Review Packet Free and Paid eBooks
 - 2nd Semester Biology Final Exam Review Packet Public Domain eBooks
 - 2nd Semester Biology Final Exam Review Packet eBook Subscription Services
 - 2nd Semester Biology Final Exam Review Packet Budget-Friendly Options
6. Navigating 2nd Semester Biology Final Exam Review Packet eBook Formats
 - ePub, PDF, MOBI, and More
 - 2nd Semester Biology Final Exam Review Packet Compatibility with Devices
 - 2nd Semester Biology Final Exam Review Packet Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 2nd Semester Biology Final Exam Review Packet
 - Highlighting and Note-Taking 2nd Semester Biology Final Exam Review Packet
 - Interactive Elements 2nd Semester Biology Final Exam Review Packet
8. Staying Engaged with 2nd Semester Biology Final Exam Review Packet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 2nd Semester Biology Final Exam Review Packet
9. Balancing eBooks and Physical Books 2nd Semester Biology Final Exam Review Packet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 2nd Semester Biology Final Exam Review Packet
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 2nd Semester Biology Final Exam Review Packet
 - Setting Reading Goals 2nd Semester Biology Final Exam Review Packet
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 2nd Semester Biology Final Exam Review Packet
 - Fact-Checking eBook Content of 2nd Semester Biology Final Exam Review Packet
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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