

## 14.1

## HYDROGEN IONS AND ACIDITY

## Section Overview

## Objectives

1. Identify conjugate acid-base pairs of ions, acids, or hydroxides and of aqueous concentrations.
2. Calculate hydroxide and hydronium ion concentrations and pH values and vice versa for a given pH.
3. Calculate the pH of weak acid solutions.

## Vocabulary

- |                   |                             |                  |
|-------------------|-----------------------------|------------------|
| 1. autoionization | 2. conjugate acid-base pair | 3. acid solution |
| 4. hydroxide ion  | 5. hydronium ion            | 6. hydronium ion |
|                   | 7. weak acid                | 8. pH            |

## Key Equations

- |   |   |
|---|---|
| 1. $\text{pH} = -\log[\text{H}_3\text{O}^+]$ (Eq. 14.1) | 2. $\text{pH} = -\log[\text{H}_3\text{O}^+]$ (Eq. 14.2) |
| 3. $\text{pH} = -\log[\text{H}_3\text{O}^+]$ (Eq. 14.3) | 4. $\text{pH} = -\log[\text{H}_3\text{O}^+]$ (Eq. 14.4) |

## Part A: Completion

Use the appropriate number to identify the relationship of the conjugate acid-base pair indicated in the second step of the acid-base reaction shown in the following equation.

1. What product is the \_\_\_\_\_ of the hydroxide ion?
2. What product is the \_\_\_\_\_ of the hydronium ion?
3. What product is the \_\_\_\_\_ of the hydroxide ion?
4. What product is the \_\_\_\_\_ of the hydronium ion?
5. What product is the \_\_\_\_\_ of the hydroxide ion?
6. What product is the \_\_\_\_\_ of the hydronium ion?
7. What product is the \_\_\_\_\_ of the hydroxide ion?
8. What product is the \_\_\_\_\_ of the hydronium ion?
9. What product is the \_\_\_\_\_ of the hydroxide ion?
10. What product is the \_\_\_\_\_ of the hydronium ion?

1. \_\_\_\_\_
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# 19 2 Hydrogen Ions And Acidity Answer Key

**Wenbin Ji**



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