

14.1

HYDROGEN IONS AND ACIDITY

Section Overview

Objectives

1. Identify conjugate acid-base pairs of ions, acids, or hydroxides and of species associated with them.
2. Calculate hydroxide ion concentrations, hydronium ion concentrations, and pH values for weak acid solutions (and vice versa).
3. Calculate the pH of weak (or strong) acids.

Vocabulary

- | | | |
|-------------------|-------------------------------|---------------------|
| 1. autoionization | 2. conjugate acid-base pair | 3. proton indicator |
| 4. autoionization | 5. hydronium ion (H_3O^+) | 6. hydronium ion |
| | 7. weak indicator | 8. pH |

Key Equations

- | | |
|---|-------------------------|
| 1. $[H_3O^+][OH^-] = 1.0 \times 10^{-14}$ (at 25°C) | 2. $pH = -\log[H_3O^+]$ |
| 3. $pH = -\log[H_3O^+]$ | 4. $pOH = -\log[OH^-]$ |

Part A: Completion

Use the appropriate number to identify the relationship of the species indicated that are associated in the reaction. Also indicate which species acts as the conjugate acid/base.

1. H_2O and H_3O^+ are _____ related. Hydrogen ion is the _____ of H_2O . The conjugate base of H_2O is _____.
2. H_2O and OH^- are _____ related. Hydroxide ion is the _____ of H_2O . The conjugate acid of H_2O is _____.
3. H_2O and H_2O are _____ related. _____ is the _____ of H_2O . _____ is the _____ of H_2O .
4. H_2O and H_2O are _____ related. _____ is the _____ of H_2O . _____ is the _____ of H_2O .
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9. H_2O and H_2O are _____ related. _____ is the _____ of H_2O . _____ is the _____ of H_2O .
10. H_2O and H_2O are _____ related. _____ is the _____ of H_2O . _____ is the _____ of H_2O .

1. _____
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19 2 Hydrogen Ions And Acidity Worksheet Answers

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Table of Contents 19 2 Hydrogen Ions And Acidity Worksheet Answers

1. Understanding the eBook 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - The Rise of Digital Reading 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Personalized Recommendations
 - 19 2 Hydrogen Ions And Acidity Worksheet Answers User Reviews and Ratings

- 19 2 Hydrogen Ions And Acidity Worksheet Answers and Bestseller Lists
- 5. Accessing 19 2 Hydrogen Ions And Acidity Worksheet Answers Free and Paid eBooks
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 - 19 2 Hydrogen Ions And Acidity Worksheet Answers eBook Subscription Services
 - 19 2 Hydrogen Ions And Acidity Worksheet Answers Budget-Friendly Options
- 6. Navigating 19 2 Hydrogen Ions And Acidity Worksheet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - 19 2 Hydrogen Ions And Acidity Worksheet Answers Compatibility with Devices
 - 19 2 Hydrogen Ions And Acidity Worksheet Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Highlighting and Note-Taking 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Interactive Elements 19 2 Hydrogen Ions And Acidity Worksheet Answers
- 8. Staying Engaged with 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 19 2 Hydrogen Ions And Acidity Worksheet Answers
- 9. Balancing eBooks and Physical Books 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 19 2 Hydrogen Ions And Acidity Worksheet Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Setting Reading Goals 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Fact-Checking eBook Content of 19 2 Hydrogen Ions And Acidity Worksheet Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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