

NAME: _____

CLASS: _____

DATE: _____

1.



An organism that eats grass and insects is

- | | | | |
|----------------------------|--------------|----------------------------|-------------|
| <input type="checkbox"/> A | a producer | <input type="checkbox"/> B | prey |
| <input type="checkbox"/> C | an herbivore | <input type="checkbox"/> D | an omnivore |

2.

Some bacteria live in the stomach of a cow. The bacteria helps the cow break down and digest the plant food that the cow eats. The bacteria gets a meal. This is an example of

- | | | | |
|----------------------------|------------|----------------------------|--------------|
| <input type="checkbox"/> A | parasitism | <input type="checkbox"/> B | Commensalism |
| <input type="checkbox"/> C | Mutualism | | |

3.



Why does every animal in an ecosystem rely on producers?

- | | | | |
|----------------------------|---------------------------------|----------------------------|---------------------------------|
| <input type="checkbox"/> A | producers eat consumers | <input type="checkbox"/> B | producers recycle |
| <input type="checkbox"/> C | producers start all food chains | <input type="checkbox"/> D | producers break down dead waste |

4.



Which organism is a decomposer?

- | | | | |
|----------------------------|------|----------------------------|-------------|
| <input type="checkbox"/> A | frog | <input type="checkbox"/> B | oak tree |
| <input type="checkbox"/> C | worm | <input type="checkbox"/> D | grasshopper |

5. Which organisms will you find the most of in an ecosystem?

Year 5 Iseb Science Test

R Sandford



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 $3 \text{3Mg} + 1 \text{Fe}_2\text{O}_3 \rightarrow 2 \text{Fe} + 3 \text{MgO}$ type of reaction single displacement
 $4 \text{1C}_2\text{H}_4 + 3 \text{O}_2 \rightarrow 2 \text{CO}_2$

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 $\text{NaOH} + \text{KNO}_3 \rightarrow \text{NaNO}_3 + \text{KOH}$ ch 2 o co 2 h 4 2 2 2o 2 fe 6 nabr 2 febr 6 na 3 caso mg
 $\text{OH} + \text{Ca(OH)}_2 + \text{MgSO}_4 \rightarrow 2 \text{NH}_4\text{OH} + \text{H}_2\text{O} + \text{NH}_4\text{Br} + \text{PbO}_2 + \text{PbO}_2 + \text{Na}_2\text{CO}_3 + \text{Na}_2\text{O} + \text{CO}_2 + \text{C}_4\text{H}_8\text{O}_4 + \text{CO}_2 + 2 \text{H}_2\text{O} + \text{HCl} + \text{NaOH} + \text{H}_2\text{O} + \text{NaCl}$

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