

Chemistry Study Guide

Periodic Table of the Elements

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Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles 🎯

Matter:

Anything that has **mass** and **occupies space** 🌟🌍

Examples: air, gases, liquids, solidz!

Atoms:

Atoms are the **smallest, indivisible particle** 🔬👉

- **Outer part** → *shell/orbit/energy level* → carries the electrons (e^-)
- **Inner part** → *nucleus* → houses the protons (p^+) and neutrons (n)
- Outer & Inner part → together these are the **subatomic particles**

Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p^+	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e^-	$\frac{1}{1836}$	-1	Orbit/Shell

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Padhraic Smyth



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web of the top 10 busiest international routes at just 160 nm 296 km the only non asian route is new york jfk jfk london heathrow lhr which is also by far the longest of the top 10 busiest international routes at 2991 nm 5563 km it is ranked 8th despite the availability of alternative routes between a

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