

SOME BASIC CONCEPTS OF CHEMISTRY

1. CHEMISTRY

Chemistry is defined as the study of the composition, properties and interaction of matter. Chemistry is often called the central science because of its role in connecting the physical sciences, which include chemistry, with the life sciences and applied sciences such as medicine and engineering.

Various branches of chemistry are

1.1 Physical chemistry

The branch of chemistry concerned with the way in which the physical properties of substances depend on and influence their chemical structure, properties, and reactions.

1.2 Inorganic chemistry

The branch of chemistry which deals with the structure, composition and behavior of inorganic compounds. All the substances other than the carbon-hydrogen compounds are classified under the group of inorganic substances.

1.3 Organic chemistry

The discipline which deals with the study of the structure, composition and the chemical properties of organic compounds is known as organic chemistry.

1.4 Biochemistry

The discipline which deals with the structure and behavior of the components of cells and the chemical processes in living beings is known as biochemistry.

1.5 Analytical chemistry

The branch of chemistry dealing with separation, identification and quantitative determination of the compositions of different substances.

2. MATTER

Matter is defined as any thing that occupies space possesses mass and the presence of which can be felt by any one or more of our five senses.

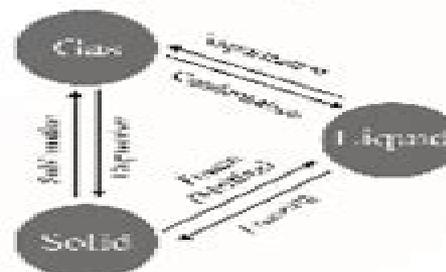
Matter can exist in 3 physical states viz. solid, liquid, gas.

Solid - a substance is said to be solid if it possesses a definite volume and a definite shape, e.g., sugar, iron, gold, wood etc.

Liquid- A substance is said to be liquid, if it possesses a definite volume but no definite shape. They take up the shape of the vessel in which they are put, e.g., water, milk, oil, mercury, alcohol etc.

Gas- a substance is said to be gaseous if it neither possesses definite volume nor a definite shape. This is because they fill up the whole vessel in which they are put, e.g., hydrogen, oxygen etc.

The three states are interconvertible by changing the conditions of temperature and pressure as follows.



3. CLASSIFICATION OF MATTER AT MACROSCOPIC LEVEL

At the macroscopic or bulk level, matter can be classified as (a) mixtures (b) pure substances.

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Reviews in Computational Chemistry, Volume 17 Kenny B. Lipkowitz, Donald B. Boyd, 2003-04-24 Computational chemistry is increasingly used in most areas of molecular science including organic inorganic medicinal biological physical and analytical chemistry Researchers in these fields who do molecular modelling need to understand and stay current with recent developments This volume like those prior to it features chapters by experts in various fields of computational chemistry Two chapters focus on molecular docking one of which relates to drug discovery and cheminformatics and the other to proteomics In addition this volume contains tutorials on spin orbit coupling and cellular automata modeling as well as an extensive bibliography of computational chemistry books FROM REVIEWS OF THE SERIES Reviews in Computational Chemistry remains the most valuable reference to methods and techniques in computational chemistry JOURNAL OF MOLECULAR GRAPHICS AND MODELLING One cannot generally do better than to try to find an appropriate article in the highly successful Reviews in Computational Chemistry The basic philosophy of the editors seems to be to help the authors produce chapters that are complete accurate clear and accessible to experimentalists in particular and other nonspecialists in general JOURNAL OF THE AMERICAN CHEMICAL SOCIETY **Computer Aids to Chemistry** Gaston Vermin, Michel Chanon, 1986 *Educational Measurement* Robert L. Brennan, 2023-10-03 Educational Measurement has been the bible in its field since the first edition was published by ACE in 1951 The importance of this fourth edition of Educational Measurement is to extensively update and extend the topics treated in the previous three editions As such the fourth edition documents progress in the field and provides critical guidance to the efforts of new generations of researchers and practitioners Edited by Robert Brennan and jointly sponsored by the American Council on Education ACE and the National Council on Measurement in Education the fourth edition provides in depth treatments of critical measurement topics and the chapter authors are acknowledged experts in their respective fields Educational measurement researchers and practitioners will find this text essential and those interested in statistics psychology business and economics should also find this work to be of very strong interest Topics covered are divided into three subject areas theory and general principles construction administration and scoring and applications The first part of the book covers the topics of validation reliability item response theory scaling and norming linking and equating test fairness and cognitive psychology Part two includes chapters on test development test administration performance assessment setting performance standards and technology in testing The final section includes chapters on second language testing testing for accountability in K 12 schools standardized assessment of individual achievement in K 12 schools higher education admissions testing monitoring educational progress licensure and certification testing and legal and ethical issues *Computing Handbook, Third Edition* Teofilo Gonzalez, Jorge Diaz-Herrera, Allen Tucker, 2014-05-07 Computing Handbook Third Edition Computer Science and Software Engineering mirrors the modern taxonomy of computer science and software engineering as described by the Association for Computing

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