

# Candlewick

The Jewel of Imperial

Price Guide  
2002  
and More...



Mary M. Wetzel-Tomalka

# Candlewick The Jewel Of Imperial Price Guide 2002 And More

**Judd E. Hollander**



**Candlewick The Jewel Of Imperial Price Guide 2002 And More:**

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web these ions are called spectator ions because they do not participate in the actual reaction canceling the spectator ions gives the net ionic equation which shows only those species that participate in the chemical reaction  $2\text{Ag}^+ \text{aq} + \text{Cr}_2\text{O}_7^{2-} \text{aq} \rightarrow \text{Ag}_2\text{Cr}_2\text{O}_7 \text{s}$  label 4 2 3

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web jan 29 2020 the ionic form of this reaction is  $2\text{Na}^+ \text{aq} + 2\text{Cl}^- \text{aq} + \text{Cu}^{2+} \text{aq} + \text{SO}_4^{2-} \text{aq} \rightarrow 2\text{Na}^+ \text{aq} + \text{SO}_4^{2-} \text{aq} + \text{CuCl}_2 \text{s}$  the sodium ions and sulfate ion are the spectator ions in this reaction they appear unchanged in both the product and reactant side of the equation

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web the six  $\text{NO}_3^- \text{aq}$  ions and the six  $\text{Na}^+ \text{aq}$  ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation  $3\text{Ba}^{2+} \text{aq} + 2\text{PO}_4^{3-} \text{aq} \rightarrow \text{Ba}_3\text{PO}_4 \text{s}$

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total ionic equations so you know in a case chemical reactions are described with chemical equations which symbolically  
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oxidation number or composition during a reaction products

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water h aq oh aq h  
2 o l the spectator ions in koh aq and hno 3 aq are k aq and no 3 aq how to identify and list the

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two ions are examples of spectator ions ions that do nothing in the overall course of a chemical reaction

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web the new format requires students to write balanced chemical equations showing only the reacting substances excluding spectator ions for three different sets of reactants and to answer a short question requiring no calculator about each reaction the questions are intended to test the students understanding of the meaning of the reactions

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