

- 1) When properly balanced, the coefficient for hydrochloric acid (HCl) would be?  
 $\underline{\quad} \text{Al} + \underline{\quad} \text{HCl} \rightarrow \underline{\quad} \text{AlCl}_3 + \underline{\quad} \text{H}_2$
- 2) The potential energy diagram for a chemical reaction is shown below



Each interval on the axis labeled "Potential Energy (kJ)" represents 40 kilojoules. What is the heat of reaction ( $\Delta H$ )?

- 3) The general equation shown is a reaction that represents what type of reaction? \_\_\_\_\_ is this reaction endothermic or exothermic? (circle one)



- 4) Write a balanced equation for the word equation shown below:  
 Aluminum + Sulfuric acid  $\rightarrow$  Aluminum sulfate + hydrogen gas

- 5) A process that absorbs heat is: endothermic or exothermic (circle one) Would the products have more energy or less energy than the reactants?
- 6) The two beakers of boiling water shown below are on hot plates set at different settings. However, the water in both beakers is at the same temperature. How is the incoming energy from the hot plates being used by the water if it is not increasing the temperature?



# 2nd Semester Final Exam Review Science Answer

**LP Steffe**



**2nd Semester Final Exam Review Science Answer:**

## **2nd Semester Final Exam Review Science Answer** Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**2nd Semester Final Exam Review Science Answer**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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