

Points to note re. use of Ballard/Westwood Timed Arithmetic Test:

General:

- Like all tests, it has its *ads and disads*. It gives a useful snap-shot of a child's reflexive knowledge of basic tables facts.
- It is timed, some *ods* do not work well under pressure of time. A low score does not always indicate a high % of incorrect answers, but could be sums incomplete.
- Child may actually know the answers but is slow at recording.
- Does not measure potential but can help identify weaknesses and assess progress. Can also indicate readiness for various algorithms, eg long division.
- **If it is not rendered useless by being used as the basis for a regular Friday tables test, it is quite reliable and efficient.**
- Children usually want to know their score and how it compares to the end-yr target. (often this is more important to them than comparing their score with other pupils)
- History of the test: a bit hazy !! Ballard (?) devised and normed the Add/Sub. Vernon revised the norms in 1949. Later Westwood devised the Mult/Div and both the add/sub & mult/div were normed by him *et al* (1974) in England and by Broughton (1975) in Australia.
- The end-yr targets have not been standardised but seem to be quite realistic and achievable (schools could decide their own targets.) These targets were set by teachers in a mixed infants to 6th, 580 pupil school, school setting neither disadvantaged or privileged. Westwood did supply mean and critically low scores for various ages, these are perhaps useful but only as a rough guide as they were standardised quite some time ago.
- By testing at Easter, the final term could be used by children to put in a final push (and also perhaps to alert parents).
- A certificate to signify reaching the End-yr Tables Target could be designed & given out.
- It might be worthwhile to administer the copying test to see if motor/hand-eye co-ordination is a factor re low scores??

Administration:

Ballard Westwood Timed Arithmetic Test

Liying Dong



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web limiting reagents and percentage yield worksheet 1 consider the reaction $\text{I}_2\text{O}_5 + 5\text{CO} \rightarrow \text{I}_2 + 5\text{CO}_2$ a 80.0 grams of iodine v oxide I_2O_5 reacts with 28.0 grams of carbon monoxide CO determine the mass of iodine I_2 which could be produced 80 g I_2O_5 1 mol I_2O_5 1 mol I_2 xs 1 333 8 g I_2O_5 1 mol I_2O_5 28 g CO 1 mol CO

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web jun 2 2020 the percent yield of a reaction is the ratio of the actual yield to the theoretical yield multiplied by 100 to give a percentage text percent yield $\frac{\text{actual yield}}{\text{theoretical yield}} \times 100$ label 3 7 3