

NOTE: Some Low Emission LMT carburetors have pressed in pilot jets (1) that are not removable. DO NOT remove the pressed in style pilot jet, Fig. 103.

3

Remove Emulsion Tube (LMT)

Using Tool #19280, Carburetor Screwdriver, remove main carburetor emulsion tube (1), Fig. 104.

Inspect Carburetor

Inspect all components for wear, damage, cracks, or plugged openings. Replace components if any of the above conditions exist. Use only compressed air and solvents to clear plugged openings.

Inspect idle mixture needle for bent needle point or a groove in tip of needle. Replace if bent or grooved.

Inspect threaded type pilot jet and passages for debris and dirt (LMT).

High Altitude Compensation

Engines set up for low altitude operation may experience a decrease in performance if operated at high altitudes.

For Engines:

LMS 1 through LMS-15, LMS-21, 22, 24, 28, 29, 30, 36

LMS 25, 26, 27, 31, 32, 34, 35, 37 and up

If poor performance is experienced refer to Table No. 5, Page 64 or Table No. 6, Page 65, Specifications, by basic Model Number and carburetor I.D. number, and remove main air bleed jet (1), Fig. 105, or fixed main jet (2), Fig. 106, if carburetor I.D. number is not found, go to "Illustrated Parts List" by Model and Type number.

LMT

High altitude kits are available, consisting of a new pilot jet, emulsion tube and side main jet, or a new emulsion tube with main jet. See "Illustrated Parts List" by Model Series and type for correct kit part number.

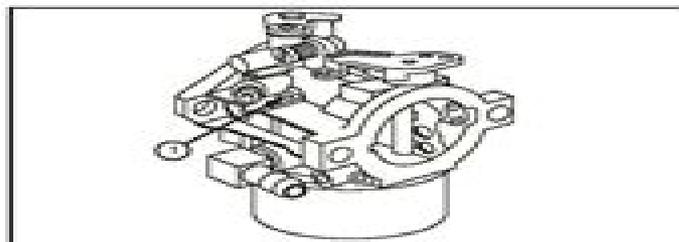


Fig. 103

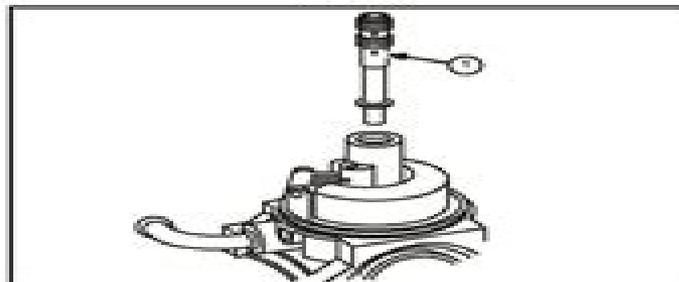


Fig. 104

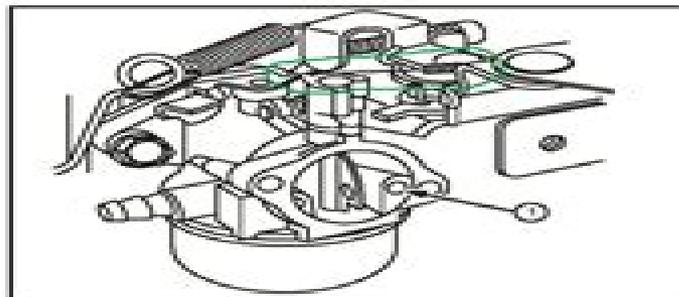


Fig. 105



Fig. 106

NOTE: SPECIFICATION TABLES BEGIN ON PAGE 53 OF THIS SECTION.

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Padhraic Smyth



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web if the force between the particles is 0.0405n what is the strength of the second charge so we rearrange the equation to solve for it what is the electric force between these two point charges are the magnitudes of the point charges is the distance between them and is a constant in this case equal to

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web choose 1 answer the solution becomes negatively charged due to the majority cl ions a the solution becomes negatively charged due to the majority cl ions the solution becomes positively charged due to the stronger mg 2 ions b the solution becomes positively charged due to the stronger mg 2 ions the solution remains neutrally

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questions time 50 minutes 40 questions directions each of the following questions or statements is followed by four possible answers choose the one best answer for each question 1

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web answer d somewhere in region d justification with the test charge and c 1 being negative there is a repulsive force on the test charge to the right from c 2 there is an attractive force on the test charge to the left by referring to coulomb s law $\square\square$
 $\square\square\square$ 1 $\square\square$ 2 $\square\square$ 2 we know that the force

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