

**BAGONG SILANG NATIONAL HIGH SCHOOL**  
**1<sup>ST</sup> QUARTER EXAMINATION**  
**PHYSICAL SCIENCE**

NAME: \_\_\_\_\_  
GRADE: \_\_\_\_\_

DATE: \_\_\_\_\_  
SCORE: \_\_\_\_\_

Directions: Write the letter of the correct answer on the space provided.

- \_\_\_\_\_ 1. It is also known as nuclear fusion and the formation of new nuclei actions in the early stages of development of the universe.  
a. Nucleosynthesis                      b. S-Process                      c. R-Process                      d. Singularity
- \_\_\_\_\_ 2-4. There are two categories of star, the average star and massive star. The life cycle of these two stars are common from 1<sup>st</sup> to 3<sup>rd</sup> stages namely nebula, protostar, and main sequence star. What will be the next stages in the life of an average star until it reaches its death?  
a. Red super giant → Supernova → Black hole → neutron star 0  
b. Red giant → white dwarf → black dwarf 3  
c. Red giant → white dwarf → black hole 2  
d. Red supergiant → white dwarf → black dwarf 1
- \_\_\_\_\_ 5. He synthesized element with atomic number 43 using a linear particle accelerator.  
a. Ernest Rutherford    b. Ernest Lawrence    c. Dmitri Mendeleev    d. John Dalton
- \_\_\_\_\_ 6. The following are light elements EXCEPT;  
a. helium                      b. hydrogen                      c. lithium                      d. iron
- \_\_\_\_\_ 7. Which of the following molecule is nonpolar?  
a. NaCl                      b. HCl                      c. CO<sub>2</sub>                      d. NH<sub>3</sub>
- \_\_\_\_\_ 8. What type of chemical bond holds the atoms of water molecule together?  
a. Hydrogen bond    b. Ionic bond    c. Polar covalent bond    d. Nonpolar covalent bond
- \_\_\_\_\_ 9. Which of the following IMFAs is considered as the weakest?  
a. H-bonding    b. Ion-dipole    c. Dipole-dipole    d. London forces
- \_\_\_\_\_ 10. Which of the following statements is TRUE about nonpolar molecules?  
a. high boiling point    b. high melting point    c. low surface tension    d. low vapor pressure
- \_\_\_\_\_ 11. Forces that are present between an ion and a polar molecule  
a. ion-dipole    b. dipole-dipole    c. dispersion forces    d. hydrogen bonding
- \_\_\_\_\_ 12. H-bonding forms when the substances involved are polar and there are \_\_\_\_\_.  
a. H-atoms of a molecule interact with to O, N, F of other molecules.  
b. C-atoms attached to O, N, F.  
c. central atoms with O, N, F as attached atoms.  
d. unshared pair of electrons in the central atom.
- \_\_\_\_\_ 13. The amount of energy required to stretch or increase the surface of a liquid by a unit area  
a. specific heat    b. surface tension    c. vapor pressure    d. heat of vaporization
- \_\_\_\_\_ 14. Liquids can form spherical elastic film to minimize surface area. What intermolecular force is responsible for the formation of this film in water?  
a. H-bonding    b. ion-induced dipole    c. dipole-induced dipole    d. London dispersion force
- \_\_\_\_\_ 15. Which group of biomolecules carries and passes on the hereditary information of the organisms?  
a. carbohydrates    b. lipids    c. nucleic acids    d. proteins
- \_\_\_\_\_ 16. What are the elements that make up carbohydrates?  
a. C and H    b. C, H and N    c. C, H and O    d. C, H, O and N
- \_\_\_\_\_ 17-19. Is carbohydrates one of the most important biological macromolecules needed by a human body?  
a. Yes, because it is the primary source of energy. 3  
b. Yes, because it can be a storage of energy for a long period of time. 1  
c. Yes, because it can be found in rice, cereals, fruits and pastas. 2  
d. Yes, because it carries genetic information. 0
- \_\_\_\_\_ 20. The measure of how fast is the change in the concentration of the reactants or products in a chemical reaction.  
a. Activation Energy    b. Activation Rate    c. Chemical Reaction    d. Concentration
- \_\_\_\_\_ 21. A chemical change that occurs when two or more substances combine to form a new substance.  
a. Activation Energy    b. Chemical Kinetics    c. Chemical Reaction    d. Product

# Section 2review Answer Key Physical Science

**Robert R. Redfield**



**Section 2review Answer Key Physical Science:**

## **Section 2review Answer Key Physical Science** Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Section 2review Answer Key Physical Science**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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web final exams reassessment of term work final exam or course marks term work and tests please refer to the instructor

faqs for any updates regarding items covered in this

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web practice final exam sta414 2104 winter 2021 university of toronto name student enrolled in course circle one sta414 sta2104 enrolled in section circle one

**final examinations academic calendar university of** - Aug 04 2023

web utsc examinations are scheduled with three start time 9 00 14 00 and 19 00 understanding the different delivery modes of final examinations in person exams

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web aug 2 2022 final exam period dec 9 21 dec 9 21 april 13 29 april 13 29 2021 summer deferred exam period dec 7 21 2021 fall deferred exam period april 9

**term work tests and final exams academic calendar** - Dec 28 2022

web examinations examinations students are expected to be available for the entire duration of the final exam period and the deferred exam period if you have submitted a petition

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web final assessments in s and y courses august 28 30 2023 deferred exam period for students who missed an in person final exam in april 2023 or june 2023 this period

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dates deadlines faculty of arts science - Dec 16 2021

web processing time it can take at least three weeks to process your request you can only request a re read for a final exam after you have paid for a reproduction request of the

final exams faculty of arts science - Oct 06 2023

web final examinations are held at the end of each academic term during the final exam period final exams are either in person or online and are scheduled at a specific time

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web on this page exam schedule conflicts students with conflicts in their final exam schedule must submit an exam conflict request via eservice during the notification period there is