

# Voltaic Cells

Label the parts for the following cell:



By connecting the two electrodes, electrons are transferred and electricity is produced.

- a. anode \_\_\_\_\_
- b. cathode \_\_\_\_\_
- c. anion \_\_\_\_\_
- d. cation \_\_\_\_\_
- e. salt bridge contains ions \_\_\_\_\_
- f. displacement cannot occur \_\_\_\_\_

- g. silver cannot displace zinc \_\_\_\_\_
- h. Zn \_\_\_\_\_

i. better metals will always displace weaker metals in a cell system. Explain by writing equations with half-reactions for copper & silver.

\_\_\_\_\_

# Voltaic Cells Study Guide

**L Reisser**

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red circular glow behind it.

## Voltaic Cells Study Guide:

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