

# Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe the components and direction of electron flow.

- a. anode \_\_\_\_\_
- b. cathode \_\_\_\_\_
- c. anion \_\_\_\_\_
- d. cation \_\_\_\_\_
- e. salt bridge (indicate flow) \_\_\_\_\_
- f. displacement (write balanced) \_\_\_\_\_
- g. half-cell reduction (write) \_\_\_\_\_
- h. Zn \_\_\_\_\_
- i. better metals will always reduce/better metals will always oxidize (written as a better argument with half-reactions for copper & zinc)

\_\_\_\_\_

# Voltaic Cells Study Guide

**S Ashworth**



## **Voltaic Cells Study Guide:**

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