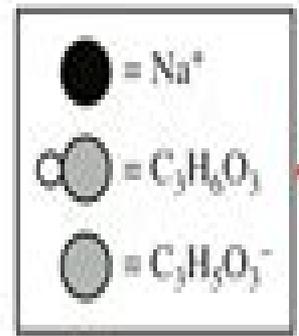


(c) Use the information in the graph to determine the approximate  $pK_a$  of lactic acid.

$$pH < pK_a$$

$$[HA] > [A^-]$$



HA  
A<sup>-</sup>

(d) The preceding diagram represents the relative amounts of major species in a sample of the solution in the flask at one point during the titration. (Note that water molecules are omitted.)

- (i) Draw an X on the preceding titration curve at a point in the titration where the reaction mixture would be represented by this diagram.
- (ii) Justify your answer.

Prior to  $\frac{1}{2}$  equiv. point



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**D Siedentop**



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