

## 16.1

## HYDROGEN IONS AND ACIDITY

## Section Overview

## Objectives

1. Identify conjugate acid-base pairs of ions, acids, or hydroxides and of aqueous concentrations.
2. Calculate hydroxide and hydronium ion concentrations and pH values and vice versa for weak acids and bases.
3. Calculate the pH of weak acid solutions.

## Vocabulary

- |                   |                                    |                   |
|-------------------|------------------------------------|-------------------|
| 1. autoionization | 2. conjugate acid-base pair        | 3. weak acid      |
| 4. hydronium      | 5. hydroxide ion ( $\text{OH}^-$ ) | 6. autoionization |
|                   | 7. weak base                       | 8. pH             |

## Key Equations

- |  |  |
|--|--|
| 1. $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$ (eq. 16.1) | 2. $\text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{OH}^-$ (eq. 16.2) |
| 3. $\text{pH} = -\log[\text{H}^+]$ (eq. 16.3)                                  | 4. $\text{pOH} = -\log[\text{OH}^-]$ (eq. 16.4)  |

## Part A: Completion

Use the appropriate number to identify the relationship of the conjugate acid-base pair indicated in the second step of the reaction. Write the number in the space provided.

1.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$  (eq. 16.1) (acid-base reaction)
2.  $\text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{OH}^-$  (eq. 16.2) (acid-base reaction)
3.  $\text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{OH}^-$  (eq. 16.2) (acid-base reaction)
4.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$  (eq. 16.1) (acid-base reaction)
5.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$  (eq. 16.1) (acid-base reaction)
6.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$  (eq. 16.1) (acid-base reaction)
7.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$  (eq. 16.1) (acid-base reaction)
8.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$  (eq. 16.1) (acid-base reaction)
9.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$  (eq. 16.1) (acid-base reaction)
10.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$  (eq. 16.1) (acid-base reaction)

1.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$  (eq. 16.1)
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9.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$  (eq. 16.1)
10.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$  (eq. 16.1)

# 19 2 Hydrogen Ions And Acidity Answer Key

**S Ben Porath**



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web tiny life ile tanışın türkiye nin önde gelen firmalarından biri olan tinylife house firması sizleri şehir hayatından uzaklaştırırken ev konforunu hissedeceğiniz dilediğiniz manzaraya göre konumlandırabileceğiniz mobil ve fonksiyonel küçük evler tasarlamaktadır

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web tiny house size dođayla i ie kk alanlarda evinizin konforunu sunmayı amalar birbirinden gzel tiny house i tasarımları yeniliki tiny house modelleri ile minimal alanları en kullanışlı şekilde planlayarak evinizi sadeleřtirir

**tiny house vega tiny house** - Apr 17 2023

web trkiye de tiny house alanında nde gelen bir firma olarak vega tiny house gnmzde konut sahibi olmak isteyen son kullanıcıya zel planlamalar yapıyor ayrıca sizlere řema oluřturma esnekliđi de veriyoruz bu sayede yeni nesil mobil tiny house tasarımlarını alternatif yařam alanlarında deneyimleme imkanı sunuyoruz