

- C. repulsion of molecules
- D. attraction of molecules
- E. surface tension of molecules.

**Answer: B**

**16. According to kinetic theory of gases, the absolute zero temperature is attained when**

- A. volume of the gas is zero
- B. pressure of the gas is zero
- C. kinetic energy of the molecules is zero
- D. specific heat of gas is zero
- E. mass is zero.

**Answer: C**

**17. Kinetic theory of gases assumes that the collisions between the molecules are**

- A. perfectly elastic
- B. perfectly inelastic
- C. partly elastic
- D. partly inelastic
- E. partly elastic and partly inelastic.

**Answer: A**

**18. The pressure of a gas in terms of its mean kinetic energy per unit volume E is equal to**

- A. E/3
- B. E/2
- C. 3E/4

**Answer: A**

**19. The pressure of a gas in terms of its mean kinetic energy per unit volume E is equal to**

- A. E/3
- B. E/2
- C. 3E/4

**Answer: A**

**20. The pressure of a gas in terms of its mean kinetic energy per unit volume E is equal to**

- A. E/3
- B. E/2
- C. 3E/4

**Answer: A**

**21. The pressure of a gas in terms of its mean kinetic energy per unit volume E is equal to**

- A. E/3
- B. E/2
- C. 3E/4

**Answer: A**

**22. The pressure of a gas in terms of its mean kinetic energy per unit volume E is equal to**

- A. E/3
- B. E/2
- C. 3E/4

**Answer: A**

**23. The pressure of a gas in terms of its mean kinetic energy per unit volume E is equal to**

- A. E/3
- B. E/2
- C. 3E/4

**Answer: A**

**24. The pressure of a gas in terms of its mean kinetic energy per unit volume E is equal to**

- A. E/3
- B. E/2
- C. 3E/4

**Answer: A**

# Multiple Choice Quizz On Thermodynamics And Fluid

**Lingjun Ying**



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and Kinetic Theory Unit X Oscillations and Waves 24 Oscillations 25 Speed of Mechanical Waves Progressive Waves 26 Superposition of Waves Interference and Beats 27 Reflection of Waves Stationary Waves in Stretched Strings and Organ Pipes 28 Doppler's Effect Log Antilog Table Value Based Questions VBO Sample Paper Examination Paper Chemical Engineering Rajaram K. Prabhudesai, 2004 Chemical Engineering Sample Exams offers the most complete set of sample exams available with step by step solutions to every problem in the book It is a superb reference guide and it provides ample practice for the exams including the new breadth depth exams *Engineering Fundamentals for Professional Engineers' Examinations* Lloyd M. Polentz, 1987 **Engineer-in-training Examination** James Warner Morrison, 1976 **Physics** Eugene Hecht, 1996 While the text covers the standard range of material from kinematics to quantum physics Hecht has carefully limited the math required to basic calculus and very basic vector analysis He omits obscure high level topics while focusing on helping students understand the fundamental concepts of modern day physics Calculus and vector analysis are both painstakingly developed as tools and then used only insofar as they illuminate the physics Hecht deliberately goes slowly justifies where each topic is going stops to take stock of where the students have been and points out the marvelous unity of the discourse Informed by a 20th century perspective and a commitment to providing a conceptual overview of the discipline this book is a return to basics **Chemical Discipline-specific Review for the FE/EIT Exam** Stephanie T. Lopina, 1997 The best preparation for discipline specific FE exams 60 practice problems with full solutions Two complete simulated 4 hour discipline specific exam Covers all the topics for that particular discipline Provides the in depth review you need Topics covered Chemical Reaction Engineering Chemical Thermodynamics Computers Numerical Methods Heat Transfer Mass Transfer Material Energy Balances Pollution Prevention Process Control Process Design Economics Evaluation Process Equipment Design Process Safety Transport Phenomena **American Journal of Physics**, 1982 **FE/EIT Mechanical Engineering Reviews** Merle C. Potter, 2001 *Industrial Discipline-specific Review for the FE/EIT Exam*, 1998 The FE exam the first in the two part engineering licensing process is taken typically by upper level students or recent graduates in April or October This eight hour exam is closed book except for a handout provided in the examination room The exam is divided into morning and afternoon sessions The morning exam with 120 multiple choice problems is the same for everyone In the afternoon examinees must choose to take a discipline specific DS or a general exam each with 60 multiple choice problems The Discipline Specific Reviews are used to study for the afternoon DS exams **Engineering Fundamentals** James Warner Morrison, 1978 *Transactions of the American Institute of Chemical Engineers*, 1911 **EIT, a Strategy for the Engineer-in-training Exam** Morton S. Fine, 1984 **Licensing Teachers** Arthur E. Wise, Rand Corporation, 1987 During the 1980s public attention to the quality of education has increased policymakers concern about the quality of teachers Although it is relatively easy to assess basic skills and academic knowledge evaluating teaching skills is far more complex and less well understood This report presents a new approach to appraising teacher

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