

JUDITH SUMMERS

Author of the bestselling MY LIFE WITH GEORGE

Who Gets Fluffy?



One collapsed marriage, one bitter divorce, one scruffy man ...



Who Gets Fluffy

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1/273 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1/273 and concluded that the pressure would approach zero at 1/273 degrees c

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web the volume of 1.00 mol of any gas at stp standard temperature 273.15 k and pressure 1 atm is measured to be 22.414 l we can substitute 101.325 kpa for pressure 22.414 l for volume and 273.15 k for temperature into the ideal gas equation and solve for r

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273.15 k and 1 atm 101.325 kpa 1 at stp one mole of an ideal gas has a volume of about 22.4 l this is referred to as the standard molar volume figure 9 18

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introduction the ideal gas law is a commonly used formula in chemistry it relates pressure temperature and volume this
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declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95
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the volume and the number of particles in each flask indicates relative amounts