

# General Chem Lab L 101 Study Guide

## Chapters 10-12

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### Chapter 10

#### *Key Concepts:*

- The types of chemical reactions: synthesis, decomposition, single replacement, double replacement, and combustion.*
- The reactants and products in each type of reaction.*
- The importance of balancing chemical equations.*
- The concept of a limiting reactant and how it can affect the outcome of a chemical reaction.*

#### *Important Events:*

- The discovery of the types of chemical reactions and the reactants and products involved in each.*
- The development of the concept of balancing chemical equations.*
- The realization of the importance of the limiting reactant in chemical reactions.*

#### *Notable Figures:*

- The scientists who discovered the types of chemical reactions and the reactants and products involved in each.*
- The scientists who developed the concept of balancing chemical equations.*
- The scientists who realized the importance of the limiting reactant in chemical reactions.*

### Chapter 11

#### *Key Concepts:*

- 1. The mole concept is important in General Chemistry because it allows for the conversion between the mass of a substance and the number of particles present. The mole is defined as the number of particles in 12 grams of carbon-12.*
- 2. The molar mass of a substance is the mass of one mole of the substance. It is important to know the molar mass of a substance in order to convert between the mass and number of*

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**S Marginson**



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