

# Voltaic Cells

Label the parts for the following cell:



By connecting the two electrodes, electrons are transferred and electricity is produced.

- a. anode
- b. cathode
- c. anion
- d. cation
- e. zinc-silver electrode
- f. silver-silver electrode

- g. zinc-silver electrode
- h. Zn
- i. Ag

2. Write the half-cell reactions for the cell shown in the diagram above. Be sure to include the correct number of electrons.

\_\_\_\_\_

\_\_\_\_\_

# Voltaic Cells Study Guide

**Marcel A. Müller**



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