

Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe a spontaneous redox reaction. (10 marks)

- a. metals _____
- b. metal ions _____
- c. metals _____
- d. metal ions _____
- e. electron movement flow _____
- f. displacement overall movement _____

- g. half cell reduction equation _____
- h. Zn _____

i. Write overall cell equation with phases. $\text{Zn(s)} + \text{Cu}^{2+}(\text{aq}) \rightarrow \text{Zn}^{2+}(\text{aq}) + \text{Cu(s)}$ (10 marks)

Voltaic Cells Study Guide

William H. Myers, William T. Mooney

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