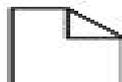


C12: Chemical Analysis



Fold page here

1. What can flame tests be used for?
2. What metal ions can flame tests be used for?
3. What colour flame do lithium compounds produce?
4. What colour flame do sodium compounds produce?
5. What colour flame do potassium compounds produce?
6. What colour flame do calcium compounds produce?
7. What colour flame do copper compounds produce?
8. Why can you not use a flame test to identify a mixture of metal ions?
9. What can sodium hydroxide be used for?
10. What metal ions can sodium hydroxide be used to identify?
11. What happens when sodium hydroxide is added to a solution containing aluminium ions?
12. What happens when sodium hydroxide is added to a solution containing calcium ions?
13. What happens when sodium hydroxide is added to a solution containing magnesium ions?
14. What happens when sodium hydroxide is added

1. Identify metal ions
2. Lithium, sodium, potassium, calcium and copper
3. Crimson red
4. Yellow
5. Lilac
6. Orange-Red
7. Green
8. Some flame colours can be masked by others
9. Identify some metal ions
10. Aluminium, calcium, magnesium, copper (II), iron (II) and iron (III)
11. A white precipitate is formed, the precipitate dissolves when an excess of sodium hydroxide is added.
12. A white precipitate is formed
13. A white precipitate is formed
14. Blue precipitate is formed.

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