

Formulas / naming of chemical compounds

Chemical compounds

Organic compounds contain carbon, usually in combination with elements such as hydrogen, oxygen, nitrogen, and sulfur.

Example: Glucose $C_6H_{12}O_6$
You learn then in organic Chemistry, IUPAC naming

Inorganic compounds... All other compounds
Exception: carbon monoxide (CO), carbon dioxide (CO_2), carbon disulfide (CS_2), compounds containing the cyanide group (CN), and carbonate (CO_3^{2-}) and bicarbonate (HCO_3^-) groups.

Ionic compounds

Metal/ non metal

Example: Sodium chloride $NaCl$

Molecular compounds

Non metal/ non metal

Example: Nitrogen dioxide NO_2

You need to learn:

1. How to find the formula from the name.
2. How to find the name from the formula.

For both molecular and ionic compounds

Hydrates

Naming not included

this semester

Acids and bases

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In a global inundated with screens and the cacophony of instant interaction, the profound power and mental resonance of verbal art frequently fade in to obscurity, eclipsed by the continuous assault of sound and distractions. However, situated within the lyrical pages of **Custom Published Dod Belleville Comm College Ss Chem 160**, a fascinating work of fictional beauty that pulses with raw feelings, lies an unique journey waiting to be embarked upon. Composed by a virtuoso wordsmith, this magical opus instructions visitors on a mental odyssey, lightly exposing the latent potential and profound influence stuck within the complicated internet of language. Within the heart-wrenching expanse of this evocative evaluation, we shall embark upon an introspective exploration of the book is main styles, dissect its interesting writing design, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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