

Chemistry Reference Sheet

Cations

| Cation name | Formula |
|-------------|------------------|
| Sodium | Na ⁺ |
| Potassium | K ⁺ |
| Magnesium | Mg ²⁺ |
| Calcium | Ca ²⁺ |
| Iron (II) | Fe ²⁺ |
| Iron (III) | Fe ³⁺ |

| Cation name | Formula |
|-------------|------------------------------|
| Hydrogen | H ⁺ |
| Copper (I) | Cu ⁺ |
| Copper (II) | Cu ²⁺ |
| Zinc (II) | Zn ²⁺ |
| Aluminum | Al ³⁺ |
| Ammonium | NH ₄ ⁺ |

Anions

| Anion name | Formula |
|------------|-----------------------------|
| Fluoride | F ⁻ |
| Chloride | Cl ⁻ |
| Bromide | Br ⁻ |
| Oxide | O ²⁻ |
| Sulfide | S ²⁻ |
| Selenide | Se ²⁻ |
| Phosphate | P ₃ ⁻ |
| Carbonate | C ₃ ⁻ |

| Anion name | Formula |
|------------|----------------------------------|
| Nitride | N ³⁻ |
| Cyanide | CN ⁻ |
| Acetate | CH ₃ COO ⁻ |
| Bromate | BrO ₃ ⁻ |
| Nitrate | NO ₃ ⁻ |
| Sulfate | SO ₄ ²⁻ |
| Carbonate | CO ₃ ²⁻ |
| phosphate | PO ₄ ³⁻ |

Covalent prefixes

| Number of atoms of an element | Prefix |
|-------------------------------|--------|
| 1 | mono- |
| 2 | di- |
| 3 | tri- |
| 4 | tetra- |
| 5 | penta- |
| 6 | hexa- |
| 7 | hepta- |
| 8 | octa- |

*the prefix for one atom is only used for the second element

Common Compounds

| Anion name | Formula |
|-------------------|----------------------------------|
| Ammonia | NH ₃ |
| Carbon dioxide | CO ₂ |
| Ethanol | C ₂ H ₅ OH |
| Hydrochloric acid | HCl |
| Hydrogen peroxide | H ₂ O ₂ |
| Nitric acid | HNO ₃ |
| Octane | C ₈ H ₁₈ |
| Sodium hydroxide | NaOH |
| Sulfuric acid | H ₂ SO ₄ |
| Water | H ₂ O |

General Equations for Calculations

Density $d = \frac{m}{V}$ Heat energy $Q = mc\Delta T$

Error $\text{error} = \text{measured} - \text{theoretical}$

%Error $\text{\%error} = \frac{\text{measured} - \text{theoretical}}{\text{theoretical}} \times 100$

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