

## FUNDAMENTALS

- A.2** Statements (c), (d), and (e) are best described as hypotheses. These statements have detected patterns: The universe has no boundaries, all animals on earth do need water to survive, and global temperature is rising, and we are generating a lot of CO<sub>2</sub> by burning coal.
- A.4** The transparency and boiling point are physical properties. The flammability of acetone is a chemical property.
- A.6** The red-brown color of copper is a physical property. The formation of copper from copper sulfide ores, the formation of copper oxide from copper, the formation of impure copper from copper oxide and carbon, and the purification of copper by electrolysis are chemical properties.
- A.8** The container (a) in Exercise A.5 shows a substance that could be a gas based on the distances between two molecules and their position in the container.
- A.10** (a) Extensive, (b) extensive, (c) extensive, (d) extensive
- A.12** (a)  $1 \times 10^{12} \text{ stadium} \times \left( \frac{10^{12} \text{ picostadium}}{1 \text{ stadium}} \right) = 1 \text{ picostadium}$
- (b)  $1 \times 10^2 \text{ pertica} \times \left( \frac{1 \text{ hectopertica}}{100 \text{ pertica}} \right) = 1 \text{ hectopertica}$
- (c)  $1 \times 10^{-1} \text{ gradus (or step)} \times \left( \frac{10 \text{ decigradus (or step)}}{1 \text{ gradus}} \right) = 1 \text{ decigradus}$

# Custom Published Chem Principles In The Lab 7e

**ML Morrison**



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