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web log [H⁺] = 10^{-4.05} = 8.91 × 10⁻⁵ mol dm⁻³ log pK_a = pH - 3.75 = 4.05 - 0.30 = 10.0 - 3.0 = 7.0 and HCOOH = 0.1 so HCOO⁻ = 0.200 mol dm⁻³ amount

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