

Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe the components and direction of electron flow.

- a. anode _____
- b. cathode _____
- c. anion _____
- d. cation _____
- e. salt bridge (indicate flow) _____
- f. displacement (write balanced) _____
- g. half-cell reduction (write) _____
- h. Zn _____
- i. better metals will always reduce/better metals will always oxidize (written as a better argument with half-reactions for copper & zinc)

Voltaic Cells Study Guide

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