



1.

1) The amino acid sequence of cytochrome *c* is exactly the same in horses and chimpanzees. There is a difference of 13 amino acids between the cytochrome *c* of humans and dogs, and a difference of 212 amino acids between the cytochrome *c* of humans and ratle snakes. Which of the following is best supported by these data?

- a. Ratle snakes apparently gave rise evolutionarily to the dog, chimpanzee, and human.
- b. Cytochrome *c* apparently has an entirely different function in ratle snakes than in mammals, which explains the difference in the number of amino acids.
- c. Cytochrome *c* is not found universally in animals.
- d. Cytochrome *c* from a ratle snake could function in a dog, but not in a chimpanzee.
- e. The human is apparently more closely related to the chimpanzee than to the dog or ratle snake.

2.

2) A parasite may be described as an organism that derives nourishment from another living organism and is therefore necessary to the life of that host.

- a. is located in the effect on the host
- b. benefits the host nutritionally
- c. provides a supportive structure for the host
- d. harms the host in the process

3.

3) Countercurrent exchange in the gills of fish is an adaptation that accomplishes which of the following?

- a. it limits the amount of blood going to the gills.
- b. it decreases the thermal gradient between water and blood.
- c. it improves the ability of blood to dissolve oxygen.
- d. it maximizes oxygen transfer from water to the blood.
- e. it maximizes heat retention in the fish.

4.

4) The energy required to run the Calvin cycle reactions of photosynthesis comes from which two substances produced during the light-dependent reactions?

- a. ATP and NADPH
- b. ADP and PCl

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**Annelies Wilder-Smith**



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web the specific heat is numerically equal to the amount of heat necessary to change the temperature of 1 00 kg of mass by 1 00 oc the si unit for specific heat is j kg

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web in order to quantify the heat transfer  $q$  that is associated with changing the temperature of a substance the mass  $m$  and specific heat  $c$  of that substance must be multiplied by

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web use this calculator to determine the heat transferred to or from a substance  $q = c\Delta t$  in joules j kilojoules kj british thermal units btu iso calories cal or kilocalories kcal

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conduction also known as thermal conduction is the transfer of internal energy within a material the energy is transferred by collision of molecules atoms and electrons inside the material the heat flow will occur in solid liquid and plasma phases and the energy will always flow from hot to cold joseph see more

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web if the reaction releases heat  $q_{rxn} < 0$  then heat is absorbed by the calorimeter  $q_{calorimeter} > 0$  and its temperature increases conversely if the reaction absorbs heat

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