

5/09/17

Gaseous State

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Q

There are 11 gaseous -

H				He
	N	O	F	Ne
			Ar	Ar
			Kr	Kr
			Xe	Xe
			Rn	Rn

Parameters of gases :-

1. **Mass** :- It is represented in form of no. of moles.

$$n = \frac{W}{M}$$

2. **Volume** :- Volume of gas = Volume of container
 $1 \text{ m}^3 = 10^3 \text{ L} = 10^3 \text{ dm}^3 = 10^6 \text{ cc} = 10^6 \text{ ml}$

3. **Pressure** :- Force exerted by gaseous molecules on the wall of container.

$$P = \frac{F}{A} \quad \text{N/m}^2 \text{ or Pa}$$

Temp. \uparrow K.E. \uparrow P \uparrow

T \propto P

$$760 \text{ mm of Hg} = 76 \text{ cm of Hg} = 1 \text{ atm} = 1.01325 \times 10^5 \text{ N/m}^2$$

$$= 1.01325 \times 10^5 \text{ Pa}$$

$$= 1.01325 \text{ bar}$$

4. **Temperature** :- Degree of hotness.

$$T(\text{K}) = T^\circ(\text{C}) + 273.15 \text{ K}$$

$$T(^{\circ}\text{F}) = \frac{9}{5} ^{\circ}\text{C} + 32$$

Chemistry Of Gaseous State

Shasha Hu



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