

4. Methanol may be produced by the exothermic reaction of carbon monoxide gas and hydrogen gas.



- (a) State the equilibrium constant expression,  $K_c$ , for the production of methanol. [1]

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- (b) State and explain the effect of changing the following conditions on the amount of methanol present at equilibrium:

- (i) increasing the temperature of the reaction at constant pressure. [2]

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- (ii) increasing the pressure of the reaction at constant temperature. [2]

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