

## 16.1

## HYDROGEN IONS AND ACIDITY

## Section Overview

## Objectives

1. Calculate concentrations based on the  $K_a$  of weak acids and hydrolysis of weak bases and equilibrium concentrations
2. Calculate  $K_a$  and  $K_b$  concentrations from  $K_a$  and  $K_b$  values and vice versa and  $K_a$  and  $K_b$  values from  $pK_a$  and  $pK_b$  values
3. Calculate  $pH$  and  $pOH$  (all solutions)

## Vocabulary

- |                   |                             |                           |
|-------------------|-----------------------------|---------------------------|
| 1. autoionization | 2. conjugate acid-base pair | 3. weak acid/base         |
| 4. autoionization | 5. hydrolysis               | 6. $K_a$ and $K_b$ values |
|                   | 7. $pH$ and $pOH$           |                           |

## Key Equations

- |   |  |
|---|--|
| 1. $K_a = \frac{[H^+][A^-]}{[HA]}$ (Eq. 16.1) | 2. $K_b = \frac{[OH^-][BH^+]}{[B]}$ (Eq. 16.2) |
| 3. $pH = -\log[H^+]$ (Eq. 16.3)               | 4. $pOH = -\log[OH^-]$ (Eq. 16.4)              |

## Part A: Completion

Use the appropriate number of significant figures in the answers to the following questions that are based on the  $K_a$  and  $K_b$  values given in the table below. Complete the answers.

1. What products are **1**  $K_a$  values of weak acids and  $K_b$  values of weak bases? The relationship of these values is given in **2**  $K_a$  and  $K_b$  values.
2. Hydrochloric acid has a  $K_a$  value of **3**  $1.3 \times 10^{-2}$ . What is **4**  $K_b$  value of its conjugate base, the chloride ion? What is **5**  $pK_a$  value of  $HCl$ ? What is **6**  $pK_b$  value of  $Cl^-$ ?
3. **7**  $K_a$  values are used to describe acids and **8**  $K_b$  values are used to describe bases. **9**  $K_a$  values are used to describe acids and **10**  $K_b$  values are used to describe bases.

1. **conjugate base**
2.  **$K_a$  and  $K_b$  values**
3.  **$1.3 \times 10^{-2}$**
4.  **$6.9 \times 10^{-3}$**
5. **1.89**
6. **2.11**
7. **acidic**
8. **basic**
9. **acidic**
10. **basic**

# 19 2 Hydrogen Ions And Acidity Answer Key

**Katrin Zwirgmaier**



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## **Table of Contents 19 2 Hydrogen Ions And Acidity Answer Key**

1. Understanding the eBook 19 2 Hydrogen Ions And Acidity Answer Key
  - The Rise of Digital Reading 19 2 Hydrogen Ions And Acidity Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying 19 2 Hydrogen Ions And Acidity Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an 19 2 Hydrogen Ions And Acidity Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from 19 2 Hydrogen Ions And Acidity Answer Key
  - Personalized Recommendations
  - 19 2 Hydrogen Ions And Acidity Answer Key User Reviews and Ratings
  - 19 2 Hydrogen Ions And Acidity Answer Key and Bestseller Lists
5. Accessing 19 2 Hydrogen Ions And Acidity Answer Key Free and Paid eBooks
  - 19 2 Hydrogen Ions And Acidity Answer Key Public Domain eBooks
  - 19 2 Hydrogen Ions And Acidity Answer Key eBook Subscription Services

- 19 2 Hydrogen Ions And Acidity Answer Key Budget-Friendly Options
- 6. Navigating 19 2 Hydrogen Ions And Acidity Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - 19 2 Hydrogen Ions And Acidity Answer Key Compatibility with Devices
  - 19 2 Hydrogen Ions And Acidity Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of 19 2 Hydrogen Ions And Acidity Answer Key
  - Highlighting and Note-Taking 19 2 Hydrogen Ions And Acidity Answer Key
  - Interactive Elements 19 2 Hydrogen Ions And Acidity Answer Key
- 8. Staying Engaged with 19 2 Hydrogen Ions And Acidity Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers 19 2 Hydrogen Ions And Acidity Answer Key
- 9. Balancing eBooks and Physical Books 19 2 Hydrogen Ions And Acidity Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection 19 2 Hydrogen Ions And Acidity Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine 19 2 Hydrogen Ions And Acidity Answer Key
  - Setting Reading Goals 19 2 Hydrogen Ions And Acidity Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 19 2 Hydrogen Ions And Acidity Answer Key
  - Fact-Checking eBook Content of 19 2 Hydrogen Ions And Acidity Answer Key
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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