

17.1

HYDROGEN IONS AND ACIDITY

Section Overview

Objectives

1. Identify conjugate acid-base pairs of ions, acids, or hydroxides and of aqueous concentrations.
2. Calculate hydroxide and hydronium ion concentrations and pH values and vice versa for weak acids and bases.
3. Calculate the pH of weak acid solutions.

Vocabulary

- | | | |
|-------------------|------------------------------------|-------------------|
| 1. autoionization | 2. conjugate acid-base pair | 3. weak acid |
| 4. hydronium | 5. hydroxide ion (OH^-) | 6. autoionization |
| | 7. weak base | 8. pH |

Key Equations

- | | |
|--|---|
| 1. $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$ (17.1) | 2. $\text{H}_2\text{O} + \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{OH}^-$ (17.2) |
| 3. $\text{pH} = -\log[\text{H}^+]$ (17.3) | 4. $\text{pOH} = -\log[\text{OH}^-]$ (17.4) |

Part A: Completion

Use the appropriate number to identify the relationship of the conjugate acid-base pair indicated in the second step of the acid-base reaction shown in the following equation.

- _____ When hydrochloric acid _____ reacts with hydroxide ion, the hydrochloride ion _____ is the conjugate base of the acid which _____ produced in this reaction. _____
- _____ Hydrochloric acid _____ reacts with _____ hydroxide ion _____ to produce _____ water _____ and _____ chloride ion _____.
- _____ Hydrochloric acid _____ reacts with _____ hydroxide ion _____ to produce _____ water _____ and _____ chloride ion _____.
- _____ _____ reacts to form two _____ water molecules and _____ hydrochloric acid _____.
- _____ Hydrochloric acid _____ reacts with _____ hydroxide ion _____ to produce _____ water _____ and _____ chloride ion _____.

- | | |
|-----|-------|
| 1. | _____ |
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19 2 Section Review Hydrogen Ions And Acidity

Judd E. Hollander



19 2 Section Review Hydrogen Ions And Acidity:

Unveiling the Energy of Verbal Beauty: An Psychological Sojourn through **19 2 Section Review Hydrogen Ions And Acidity**

In a world inundated with displays and the cacophony of instantaneous connection, the profound power and psychological resonance of verbal art often diminish into obscurity, eclipsed by the constant onslaught of sound and distractions. However, set within the musical pages of **19 2 Section Review Hydrogen Ions And Acidity**, a charming work of fictional splendor that pulses with natural emotions, lies an wonderful journey waiting to be embarked upon. Composed with a virtuoso wordsmith, this enchanting opus courses visitors on an emotional odyssey, gently revealing the latent possible and profound influence stuck within the intricate web of language. Within the heart-wrenching expanse with this evocative evaluation, we shall embark upon an introspective exploration of the book is main styles, dissect its captivating writing style, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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