

# SETTING UP A NETWORK PRINTER.

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# Network Printer Manual

**James Pyles**



## **Network Printer Manual:**

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web the molar volume has the si unit of cubic metres per mole m<sup>3</sup> mol although it is more typical to use the units cubic decimetres per mole dm<sup>3</sup> mol for gases and cubic centimetres per mole cm<sup>3</sup> mol for liquids and solids

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standard pressure the volume occupied by one mole of a gas is called molar volume therefore the molar volume of a gas at standard temperature and pressure stp is 22.4 l

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web at standard temperature and pressure stp the molar volume  $V_m$  is the volume occupied by one mole of a chemical element or a chemical compound it can be calculated by dividing the molar mass  $M$  by mass density  $\rho$  molar gas volume is one mole of any gas at a specific temperature and pressure has a fixed volume molar volume formula

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web jun 8 2023  $v = 1.00 \text{ mol} \times 0.08206 \text{ latm mol}^{-1} \text{ k}^{-1} \times 273 \text{ k} = 22.4 \text{ l}$  in other words 1 mole of a gas will occupy 22.4 l at stp

assuming ideal gas behavior at stp the volume of a gas is only dependent on number of moles of

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