

Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe a spontaneous redox reaction. (10 marks)

- a. Anode: _____
- b. Cathode: _____
- c. Anion: _____
- d. Cation: _____
- e. Half-cell reactions: Zn: _____
Cu: _____
- f. Overall balanced redox reaction: _____
- g. Cell notation: _____
- h. Write a half-cell with a zinc electrode in a zinc sulfate solution and a silver electrode in a silver sulfate solution with a salt bridge with a voltmeter. (10 marks)

Voltaic Cells Study Guide

Julie R. Frentrup



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