

Voltaic Cells

Label the parts for the following cell:



By connecting the two electrodes, electrons can be transferred and electricity is produced.

- a. anode _____
- b. cathode _____
- c. anion _____
- d. cation _____
- e. salt bridge _____
- f. direction of electron flow _____
- g. direction of ion movement _____

- g. silver-silver chloride system _____
- h. Zn _____

2. Write the half-cell and overall cell reactions for the cell represented by the notation $\text{Zn} | \text{Zn}^{2+} || \text{Ag}^{+} | \text{Ag}$.

Voltaic Cells Study Guide

Ian Pickup



Voltaic Cells Study Guide:

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