
What Is Geochemistry?

Geology began when early man first picked up a stone, considered its qualities, and decided that it was better than the stone he already had. Good stones were useful and they were collected, mined, and traded.

HENRY FAUL AND CAROL FAUL (1963, p. 1)

Geochemistry is based on the urge to understand why some stones are “good” and how they formed. It involves applications of the principles of chemistry to the solution of geological problems and therefore could not develop until chemistry and geology had been established as scientific disciplines. These applications were practiced first during the 16th century in the mines of Europe where knowledge of minerals and their chemical compositions was used to recognize and follow veins of ore. For example, in 1574 Lazarus Ercker, the superintendent of mines for King Rudolf II of Austria, published a manual in which he described ores and outlined procedures for their analysis (Ercker, 1951). However, several centuries passed before geochemistry in the present sense of the word became established as an integral component of Earth and Planetary Science.

1.1 Early History

Two essential prerequisites for the growth of geochemistry were the discovery of the chemical elements and the development of sensitive and accurate methods for the analysis of rocks and minerals. Until the end of the 16th century matter was thought to consist of earth, water, fire, and air, whose basic qualities were described as

warm, cold, dry, and wet. The metals gold, silver, copper, iron, tin, mercury, and lead together with sulfur and carbon had been known for thousands of years but were not recognized as elements in the present sense of the word. Antimony, arsenic, bismuth, and phosphorus were studied by the alchemists during the Middle Ages.

The development of analytical chemistry during the 18th century resulted in the discovery of 46 chemical elements between 1720 and 1850. After Bunsen and Kirchhoff invented the optical emission spectrograph, 30 additional elements were added from 1850 to 1925. The transuranium elements and certain other radioactive elements were discovered later in the 20th century (Correas, 1969).

As the number of known chemical elements increased during the 19th century, chemists became aware that they could be organized into groups based on similarities of their chemical properties. These tendencies became the foundation of the periodic table of the elements proposed independently by D. I. Mendeleev in 1869 and by J. L. Meyer in 1870 (Asimov, 1965). The cause for the periodicity of the chemical properties of the elements was not known then and was recognized only after the internal structure of atoms was worked out during the first 30 years of the 20th century.

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Sir Kingsley Charles Dunham



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