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## NIMS/700 Questions And Answers With Verified Study Set

Ensuring the uninterrupted flow of information describes which key communications and information systems principle?  Resilience and Redundancy

Which type of ICS facility is used to temporarily position and account for personnel, supplies, and equipment awaiting assignment?  Staging Area

Which NIMS structure makes cooperative multi-agency decisions?  MAC Groups

Each ICS General Staff is led by a \_\_\_\_\_ who reports directly to the incident Commander or Unified Command  Section Chief

The Jurisdiction receiving mutual aid can \_\_\_\_\_ resources if they do not meet its needs.  Deline

In NIMS, when do managers plan and prepare for the demobilization process?  At the same time they begin mobilizing resources.

Emergency Management Assistance Compact(EMAC)is a state-to-state system for sharing resources during an emergency or disaster.  True

Which NIMS Management Characteristic follow established processes for gathering, analyzing, assessing, sharing and managing data?  Information and Intelligence Management

ICS provides a standardized approach to the command, control, and coordination of \_\_\_\_\_ emergency personnel.  On-scene

In NIMS, resource inventory refers to preparedness activities conducted \_\_\_\_\_(of) incident response.  outside

Which NIMS command and Coordination structures are offsite locations where staff from multiple agencies come together?  Emergency Operations Centers (EOC'S)

Which NIMS Management Characteristic includes maintaining accurate and up-to-date inventories of personnel, equipment, teams,and supplies?  Comprehensive Resource Management

Which resource management task determines the type, quantity, receiving location, and users of resources?  Identify Requirements

# 700 Nims Test Answers

**K Payea**



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web background this experiment is designed to study the kinetics of a chemical reaction the reaction involves the oxidation of iodide ions by bromate ions in the presence of acid  $6\text{I}^- + \text{BrO}_3^- + 6\text{H}^+ \rightarrow 3\text{I}_2 + \text{Br}^- + 3\text{H}_2\text{O}$  reaction 1 the reaction is somewhat slow at room temperature the reaction rate depends

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