

## CHEMISTRY UNIT 1 STUDY GUIDE.

### What is Matter?

- Solid, Liquid, gas
- Anything that takes up space & has mass
- Mass is the amount of matter in an object.

### Types

- Substance- a particular kind of matter – pure- made up of elements and compounds
- Mixture- more than one kind of matter. 2+ elements and compounds

### Properties:

**Physical** – property that can be observed and measure without changing the substances.

**Chemical** – property that can only be observed by changing the type of substance.

**Extensive**- only depends on the amount of matter

Mass, weight, volume, length

**Intensive** – only depends on the type of matter, not the amount used to identify a substance.

Color, Odeur, melting/ freezing point.

**Ductility**- the ability to make copper into wires.

### States of Matter:

	Definite volume	Definite shape	Temperature increase	compressible
SOLID	Yes	Yes	Small expans	No
LIQUID	Yes	No Takes up shape of container	Small expans	No
GAS	No	No	Large expans	Yes



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**National Science Foundation (U.S.)**



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## The Enigmatic Realm of **Science Review Sheet Unit Freshman**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Science Review Sheet Unit Freshman** a literary masterpiece penned by a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of people who partake in its reading experience.

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