

Voltaic Cells

Label the parts for the following cell:



By connecting the two electrodes, electrons are transferred and electricity is produced.

- a. anode
- b. cathode
- c. anode
- d. cathode
- e. zinc anode (oxidation) half-cell
- f. silver anode (oxidation) half-cell

g. silver anode (oxidation) half-cell

h. Zn

i. better metals will always oxidize (lose electrons) and be oxidized (reduced) by better (more noble) metals (remember the activity series)

Voltaic Cells Study Guide

Robert K. Wismer



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